(-)-GERANYLLINALOOL FROM THE OLEORESIN

OF Picea obovata

N. V. Avdyukova, É. N. Shmidt, and V. A. Pentegova

UDC 547.597

We have previously reported the isolation from the high-boiling neutral faction of the oleoresin of Picea obovata Ledb. (Siberian spruce) of cembrene, isocembrol, neoabienol, dehydroabietinol, neocembrene, and epimanoyl oxide [1]. Continuing our investigation of this fraction of the oleoresin, we have isolated a compound with the empirical formula $C_{20}H_{34}O$ (n_D^{24} 1.4878), [α]_D -14.1° (c 0.34; chloroform), mol. wt. 290 (mass spectrometry), the hydrogenation of which in acetic acid in the presence of PtO₂ showed the presence of four double bonds in it. The UV spectrum of the compound lacks a maximum in the 220-240 nm region.

The IR spectrum contains absorption bands characteristic for a hydroxy group (3615, 1120 cm⁻¹), a vinyl group (930, 1000, 1655, 3090 cm⁻¹), and a trisubstituted double bond (850, 1680 cm⁻¹).

The NMR spectrum (CCl₄, 60 MHz) has the signals of four methyl groups at 1.56 ppm (9H) and 1.65 ppm (3H), these being present on double bonds. A singlet with a chemical shift of 1.20 ppm is assigned to a meth-

yl in a C fragment. In the low-field region a signal at 5.00 ppm with an intensity of 3H is due to the olefinic protons of a C bond, and multiplets with chemical shifts of 5.80, 5.18, and 4.86 ppm are characteristic for the protons of a vinyl double bond.

The facts above enable the compound isolated to be identified as geranyllinalool (I) [2].

Geranyllinalool, like geranylgeraniol, is considered a possible biogenetic precursor of the cyclic diterpenoids [3]. Its presence in a natural material rich in diterpenes may be considered as an indirect confirmation of Ruzicka's hypothesis of the formation of bi- and tricyclic diterpenes from an acyclic compound [4].

In addition to geranyllinalool, we isolated a norditerpene alcohol of the dehydroabietane type with mp 91-92°C, $[\alpha]_D$ +53.3° (c 0.34; methanol) with spectral characteristics and a melting point close to those for 18-nordehydroabietan-4- α -ol [5, 6] and to the hydrocarbon 18-nordehydroabiet-4(19)-ene, with n_D^{24} 1.5300, $[\alpha]_D$ +154° (c 0.40; chloroform). Literature data: $[\alpha]_D$ +152° (c 0.20; methanol) [6].

LITERATURE CITED

- 1. É. N. Shmidt and V. A. Pentegova, Khim. Prirodn. Soedin., 6, 769 (1970).
- 2. B. Kimland and T. Norin, Acta Chem. Scand., 21, 825 (1967).

Novosibirsk Institute of Organic Chemistry, Siberian Branch, Academy of Sciences of the USSR. Translated from Khimiya Prirodnykh Soedinenii, No. 6, pp. 839-840, November-December, 1971. Original article submitted June 22, 1971.

• 1971 Consultants Bureau, a division of Plenum Publishing Corporation, 227 West 17th Street, New York, N. Y. 10011. No part of this publication may be reproduced, stored in a retrieval system, or transmitted, in any form or by any means, electronic, mechanical, photocopying, microfilming, recording or otherwise, without written permission of the publisher. A copy of this article is available from the publisher for \$15.00.

- L. Ruzicka, Exper., 9, 357 (1953).
- 4. L. Ruzicka, Pure Appl. Chem., 6, 493 (1963).
- 5.
- R. N. Seelye and W. B. Watkins, Tetrahedron Lett., 1968, 1271. C. R. Bennett, R. C. Cambie, R. A. Franich, and T. J. Fullerton, Aust. J. Chem., 22, 1711 (1969).